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Nomenclature

a	Number of Carbon atoms
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c	Number of Oxygen atoms
d	Number of Nitrogen atoms
$\overline{c_p}$	Mole specific heat at constant pressure
$c_{p,i}$	Specific heat of species i
\overline{h}	Mole specific enthalpy
$\overline{h}_{f,i}^0$	Enthalpy of formation
H_R	Enthalpy of reactants
H_P	Enthalpy of products
K	Equilibrium constant
m_i	Mass of species i
M_i	Molecular weight of species i
n	Total number of moles
n_i	Number of moles of species i
$n_{R,i}$	Number of moles of species i for reactants
$n_{P,i}$	Number of moles of species i for products
P	Total pressure
P_1, P_2	Total pressure at states 1 and 2
Q	Heat loss
R	Gas constant
\overline{R}	Universal gas constant
\overline{s}^o	Mole specific entropy
T_a	Adiabatic flame temperature
T_R	Reactants temperature
T_P	Products temperature
T_{ref}	The reference temperature
U_1, U_2	Internal energy at states 1 and 2
V_1, V_2	Total volume at states 1 and 2
W	Work done
x_i	Mole fraction of species i
z_i	Number of mole of species i
Φ	Equivalence ratio
ϵ	Molar fuel air ratio

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