

# **Sudan University of Science and Technology College of Graduate Studies**

# **Development and Validation of an Assay for Bisoprolol Fumarate and Hydrochlorothiazide Multicomponent Drug**

تطوير وتحقق من طريقة لتحليل بيسوبرولول فومارات وهيدروكلوروثيازايد -عقار متعدد المواد

**A Thesis submitted in fulfillment of the requirements for a master degree in chemistry**

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**Supervisor Dr. Mohamed Suleiman Ali May - 2018**

الآيــــــة

قال تعالـــــــ*ي*:

﴿ يَرْفَعِ اللَّهُ الَّذِينَ آَمَنُوا مِنْكُمْ وَالَّذِينَ أُوتُوا الْعِلْمَ دَرَجَاتٍ وَاللَّهُ بِمَا تَعْمَلُونَ خَبِير ﴾ **ِ سورة المجادلة آية )11(**

# **Dedication**

**Dedicate**

**To**

**My Mother**

**My father**

**My wife and daughter**

# **Acknowledgements**

**I would like to thank God for giving me health to accomplish this work. Many thanks also my supervisor Dr. Mohammed Suleiman Ali for this support, encouragement and assistance throughout the provider of research he has cheerfully answered my queries, provided me with materials, checked my result, assisted me in willingly in writing up and helpfully commented on the earlier drafts of this project.**

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#### **Abstract**

A simple, precise, accurate and economical Reverse phase HPLC with a detection UV, method was developed and validated for the simultaneous determination of Bisoprolol fumarate and Hydrochlorothiazide in pharmaceutical formulations. Better separation was achieved using Cyanide column (250mm  $\times$  4.6mm, 5µm), maintained at 30 °C. The mobile phase was composed of buffer: methanol in the ratio of 82:18 (v/v), (the buffer was 1% solution of tetra butyl ammonium hydroxide, adjusted to  $pH = 5.0$  by acetic acid). Isocratic elution was used with a flow rate of 0.9 ml/min, injection volume was 10μ, and effluents were monitored by UV (228nm). The retention time of Bisoprolol fumarate and Hydrochlorothiazide was 4.7min and 7.7min, respectively.

The method was tested for linearity over a concentration range of (40-160) μg/ml and (100-400) μg/ml, for Bisoprolol fumarate and hydrochlorothiazide, respectively; the correlation coefficient  $(R^2)$ , was found to be 0.9998 and 0.9999, respectively.

The method successfully passed all validation tests stipulated in the validation protocol of International Conference on Harmonization (ICH) and united State Pharmacopeia (USP).

#### مستخلص البحث

طريقه سهلة، دقيقة واقتصادية باستخدام كروماتوغرافيا السائل عالية الضغط مع مكشاف االشعة فوق البنفسجية، تم تطويرها والتحقق منها لتحليل خليط عقار ثنائي مكون من بيسوبرولول فومارات وهيدروكلوروثيازايد. وقد تم الفصل بصورة جيدة باستخدام عمود السيانيد )250مم \* 4.6 مم\* 5 مايكرومتر( عند درجة حرارة 30 درجة مئوي. الطور المتحرك يتكون من محلول منظم وميثانول بنسبة 18:82 على التوالي )المحلول المنظم يحتوي على %1 محلول تترا بيوتيل امونيوم هيدروكسيد، تم ضبط الأس الهيدروجيني على ٥٫٠ بواسطة حمض الخليك). وتم استخدام الازاحة احادية الطور المتحرك للفصل، وكان معدل سريان الطور المتحرك 0.9 مل لكل دقيقة، كمية حقن العينة 10مايكرولتر، وقد تم تقدير المادتين الفعالتين عند طول موجي 228 نانوميتر وكان زمن استبقاء بيسوبرولول فومارات وهيدروكلوروثيازايد 4.7 دقيقة و7.7 دقيقة على التوالي.

تمت دراسة خطية العلاقة لمادة بيسوبرولول فومارات في مدى التراكيز (٤٠ مايكرو جرام/مل -١٦٠مايكرو جرام/مل) وهيدروكلوروثيازايد في مدى التراكيز (١٠٠ مايكرو جرام/مل -٤٠٠ مايكرو جرام/مل) فكان معامل الخطية لمادة بيسوبرولول فومارات 0.9998 ولمادة هيدروكلوروثيازايد 0.9999 على التوالي.

وقد اجتازت هذه الطريقة بنجاح جميع اختبارات التحقق المنصوص عليها ببروتوكول المؤتمر الدولي المعني بالمتطلبات التقنية لتسجيل الصيدالنيات الخاصة باإلنسان ودستور الواليات المتحدة األمريكية لألدوية.

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**Chapter one Introduction and Literature Review**

# **Chapter one 1. Introduction and Literature Review**

#### **1.1 Introduction**

Hypertension is a major public health problem of worldwide distribution and is a major risk factor of cardiovascular disease morbidity and mortality. It is responsible for one-half of coronary heart disease (CHD) and about two thirds of cerebrovascular accidents. The relationship between blood pressure and risk of cardiovascular disease events is continuous, consistent and independent of other risk factors. The higher the blood pressure, the greater is the chance of myocardial infarction, heart failure, stroke and kidney disease. (Kavita, et al., 2013).

It is surprising that only about 50 years ago hypertension was considered an essential malady and not a treatable condition. Introduction of thiazide diuretics in late 50s made some headway in successful treatment of hypertension and ambitious multicenter VA cooperative study (phase 1 and 2) started in 1964 for diastolic hypertension ranging between 90 and 129 mmHg and completed by 1971 established for the first time that treating diastolic hypertension reduced CV events such as stroke and heart failure and improved mortality. In the next two decades, ALLHAT and other studies examined the comparability of outcomes with use of different classes and combinations of antihypertensive drugs. (Mohammad G. Saklayen and Neeraj V. Deshpande 2016).

Antihypertensive drugs d is a class of drugs that has an important place in the range of medicinal products currently used to treat cardiovascular diseases. Antihypertensive drugs combine several active ingredients with different mechanisms of action, but with synergistic action, have better tolerability and increased effectiveness. Although the first choice for reducing blood pressure is the lifestyle, expressed in diet and exercises, most patients also need drug therapy. The most commonly used antihypertensive drugs are diuretics, angiotensin-converting enzyme (ACE) inhibitors, angiotensin antagonists and calcium channel blockers, and in some cases, a combination of two or three of these needed. (Moisei et al., 2016).

Combining blood pressure-lowering drugs from different constituents, approximately, 5 times more effective than doubling the dose of one drug. It follows that to maximize efficacy combination therapy, preferably using low doses to minimize side effects, is substantially better than monotherapy and should be considered as routine initial therapy.

Obtaining the target blood pressure level by monotherapy can be currently challenging, especially for the patients who are suffering meanwhile from other diseases, It is demonstrated that a majority of hypertensive patients need two or more antihypertensive drugs to effectively lower their blood pressure. Consequently, fixed-dose that can be defined as that several active agents were combined in single pharmaceutical formulations appears to be a novel and underlying asset in overcoming the cardiovascular disease. Based on the analysis of some literature and relative data from FDA, the advantages of fixed-dose combination are elucidated, and formulations of common dual, triple combinations were summarized. Clinical practices proved that fixed-dose combinations had many benefits comparing with single drug and separate agents in terms of effects, convenience, compliance, and costs to a certain extent. From patients' perspective, fixeddose combination therapy will be increasingly utilized in blood pressure control in the future. (Wald et al., 2009; Xinhuan Wan et al., 2014).

#### **1.1.1 Bisoprolol Fumarate and Hydrochlorothiazide**

#### **i. Bisoprolol Fumarate**

The molecular formula for Bisoprolol fumarate is:  $(C_{18}H_{31}NO_4)$ <sub>2</sub>.  $C_4H_4O_4$ , Molecular Weight 766.98 and IUPAC name is  $(\pm)$ -1-[4-[[2-(1-methylethoxy) ethoxy] methyl] phenoxy]-3-[(1-methylethyl) amino]-2-propanol (E) -2-butenedioate (2:1) and the chemical structure is shown in Figure1.1.



Figure 1.1: **Is Bisoprolol Fumarate structure. (USP 2016)**

Bisoprolol fumarate is a cardio selective β 1 -adrenergic blocker. It possesses an asymmetric carbon atom in its structure and is provided as a racemic mixture. The S (-) enantiomer is responsible for most of the beta-blocking activity. It is, almost completely, absorbed from the gastrointestinal tract and undergoes minimal first pass metabolism resulting in an oral bioavailability of about 90%. It is bound to plasma proteins at about 30%.

Bisoprolol has been used individually, and in combination with other antihypertensive agents for the treatment of hypertension, heart attacks, and kidney problems. (Savita's Yadav and Janhavirrao .2013; Bozal et al. 2013; Raju et al. 2016; Renuka et al. 2016).

#### **ii. Hydrochlorothiazide**

The molecular formula for Hydrochlorothiazide is  $C_7H_8C1N_3O_4S_2$ , Molecular Weight (297.74) and IUPAC name is 6-Chloro-3, 4-dihydro-2H-1, 2, 4-benzothiadiazine-7 sulfomanide 1, 1-dioxide and the chemical structure is shown in Figure1.2.



Figure 1.2: **Is Hydrochlorothiazide structure. (USP 2016)**

Hydrochlorothiazide is a diuretic/antihypertensive agent. is used mainly for treatment of mild to moderate hypertension of edema in people with congestive heart failure, cirrhosis of the liver, or kidney disorders and is, usually, administered with other drugs. Hydrochlorothiazide binds to and inhibits the enzyme carbonic anhydrase.

It is, frequently, used alone or in combination with other medications for the treatment of hypertension, congestive heart failure, symptomatic edema, diabetes insipidus, renal tubular acidosis, hyperparathyroidism, and edema and prevention of kidney stones and used in the treatment of osteoporosis. (Savita's Yadav and Janhavirrao.2013; Renuka et al. 2016; Nidhal S. Mohammed and Ahmed J.Mohammed.2016).

#### **1.1.2 Analytical procedure**

The analytical procedure refers to the way of performing analysis. It describes, in details, the steps necessary to perform each ,analytical, test. This may include but is not limited to the sample, the reference standard and reagents preparation, use of the apparatus, generation of calibration curves, use of the formulae for the calculation, etc.

#### **1.1.3 Methods validation**

Methods are validated for System Suitability specificity, linearity, precision, accuracy and Robustness.

## **1.1.3.1 System Suitability**

System suitability testing is an integral part of many analytical procedures. Tests are based on the concept that the equipment, electronics, analytical operations and samples to be analyzed constitute an integral system that can be evaluated as such.

System suitability test parameters to be established for a particular procedure depend on the type of procedure being validated.

The parameters used in the system suitability tests report are as follows:

- Number of theoretical plates (Not less than 2000).
- Resolution (Not less than 2.0).
- Tailing factor (Not more than 2.0).
- Relative Standard Deviation (Not more than 2%).

## **1.1.3.2 Specificity**

Specificity is the ability to assess, unequivocally, the analyte in the presence of components, which may be expected to be present. Typically, these include impurities, degradants, matrix, etc. Lack of specificity of an individual analytical procedure may be compensated by other supporting analytical procedure. This definition has the following implications:

Identification: to ensure the identity of an analyte.

Purity Tests: to ensure that all analytical procedures performed allow an accurate statement of the content of impurities of an analyte, i.e. related substances test, heavy metals, residual solvents content, etc.

#### • **Assay (content or potency)**

To provide an exact result which allows an accurate statement on the content or potency of the analyte in a sample.

#### **1.1.3.3 Linearity and Range**

The linearity of an analytical procedure is its ability (within a given range) to obtain test results which are directly proportional to the concentration (amount) of analyte in the sample. Linearity should be evaluated by visual inspection of a plot of signals as a function of analyte concentration or content. If there is a linear relationship, test results should be evaluated by appropriate statistical methods, for example, by calculation of a regression line by the method of least squares. In some cases, to obtain linearity between assays and sample concentrations, test data may need to be subjected to a mathematical transformation prior to regression analysis. Data from the regression line itself may be helpful to provide mathematical estimates of the degree of linearity.

The range of the procedure is validated by verifying that the analytical procedure provides acceptable precision, accuracy, and linearity when applied to samples containing analyte at the extremes of the range as well as within the range.

#### **1.1.3.4 Detection Limit**

The detection limit of an individual analytical procedure is the lowest amount of analyte in a sample, which can be detected but not, necessarily quantitated as an exact value. The detection limit is, generally, expressed in the concentration of analyte (ppm) in the sample. A number of approaches are recommended by the ICH for determining the detection limit of sample, depending on instrument used for analysis, nature of analyte and suitability of the method. The acceptable approaches are:

- Visual evaluation.
- Signal-to-noise ratio.
- Standard deviation of the response.
- Standard deviation of the slope of linearity plot.

The formula for calculating LOD is:

$$
LOD = 3.3 \delta/S
$$

Where  $\delta$  = the standard deviation of the response

 $S =$  the slope of the calibration curve

## **1.1.3.5 Quantitation Limit**

The quantitation limit of an individual analytical procedure is the lowest amount of analyte, in a sample, which can be, quantitatively, determined with suitable precision and accuracy. The quantitation limit is a parameter of quantitative assays for low levels of compounds in sample matrices and is particularly used for the determination of impurities and/or degradation products. Like LOD, ICH recommends the following four methods for estimation of LOQ.

The acceptable approaches are:

- Visual evaluation.
- Signal-to-noise ratio.
- Standard deviation of the response.
- Standard deviation of the slope of linearity plot.

The formula for calculating LOQ is:

$$
LOQ = 10 \delta/S
$$

Where  $\delta$  = the standard deviation of the response.

 $S =$  the slope of the calibration curve

#### **1.1.3.6 Accuracy**

The accuracy of an analytical procedure expresses the closeness of agreement between the value, which is accepted either as a, conventional, true value or as, an accepted, reference value, and the value found. This is sometimes termed trueness.

In the case of an assay of a drug substance, accuracy may be determined by application of an analytical procedure to an analyte of known purity (e.g., a Reference Standard) or by comparison, of the results of the procedure with those of a second, well-characterized procedure, the accuracy of which has been stated or defined.

In the case of the assay of a drug in a formulated product, accuracy may be determined by application of the analytical procedure to synthetic mixtures of the drug product components to which known amounts of analyte have been added within the range of the procedure. If it is not possible to obtain samples of all drug product components, it may be acceptable either to add known quantities of the analyte to the drug product (i.e., "to spike") or to compare results with those of a second, well-established procedure, the accuracy of which has been stated or defined.

In the case of quantitative analysis of impurities, accuracy should be assessed on samples (of drug substance or drug product) spiked with known amounts of impurities. Where it is not possible to obtain samples of certain impurities or degradation products, results should be compared with those obtained by an independent procedure. In the absence of other information, it may be necessary to calculate the amount of an impurity based on comparison of its response to that of the drug substance; the ratio of the responses of equal amounts of the impurity and the drug substance (relative response factor) should be used, if known.

#### **1.1.3.7 Precision**

The precision of an analytical procedure expresses the closeness of agreement (degree of scatter) between a series of measurements obtained from multiple sampling of the same homogeneous sample under prescribed conditions. Precision may be considered at three levels: repeatability, intermediate precision and reproducibility. Precision should be investigated using homogeneous, authentic samples. However, if it is not possible to obtain a homogeneous sample it may be investigated using artificially, prepared samples or a sample solution.

The precision of an analytical procedure is, usually, expressed as the variance, standard deviation or coefficient of variation of a series of measurements.

The precision of an analytical procedure is determined by assaying a sufficient number of aliquots of a homogeneous sample to be statistically able to calculate valid estimates of standard deviation or relative standard deviation (coefficient of variation). Assays in this

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context are, independent, analyses of samples that have been carried through the complete analytical procedure from sample preparation to final test result.

The ICH documents recommend that repeatability should be assessed using a minimum of nine determinations covering a specified range of a procedure (i.e., three concentrations and three replicates of each concentration) or using a minimum of six determinations at 100% of the test concentration.

#### **1.1.3.8 Robustness**

The robustness of an analytical procedure is a measure of its capacity to remain unaffected by small, but deliberate variations in method parameters and provides an indication of its reliability during normal usage.

If measurements are susceptible to variations in analytical conditions, the analytical conditions should be, suitably, controlled or a precautionary statement should be included in the procedure. One consequence of the evaluation of robustness, should be that a series of system suitability parameters (e.g., resolution test) is established to ensure that validity of the analytical procedure, is maintained whenever used. (ICH 2005; USP 2016).

#### **1.2 Literature Review**

#### **1.2.1 HPLC methods**

Patel et al. (2006); reported HPLC-UV method the determinate for bisoprolol Fumarate and Hydrochlorothiazide was developed and validated; using  $C_{18}$  column (20 x 4.6mm, 5μm) at ambient temperature, injection volume was 20μl. The mobile phase was water, acetonitrile and tetrahydrofuran in the ratio of 80:20:5 (v/v). Isocratic elution was used with 1ml/min flow rate, the eluents were UV monitored at 225nm. The Linearity ranges of Bisoprolol Fumarate and Hydrochlorothiazide were 10-150μg/ml and 1-90μg/ml; respectively. Limits of detection of Bisoprolol Fumarate and Hydrochlorothiazide were  $3.5\mu g/ml$  and  $0.4\mu g/ml$ , while limits of quantitation were  $8.5\mu g/ml$  and  $0.9\mu g/ml$ ; respectively.

Qutab al. (2007); developed HPLC-UV method for determination of hydrochlorothiazide and candesartan Cilexetil using a Phenyl -2 column, mixture of 0.02M potassium dihydrogen phosphate, methanol and triethyl-amine, as mobile phase in the ratio of 25: 75: 0.2 (v/v), final pH  $6.0 \pm 0.1$ , as mobile phase. The flow rate was 1ml/min and eluents were monitored at 271nm and linearity ranges were 5–45µg/ml and 12–56µg/ml; respectively. The limits of detection were 0.08μg/ml and 0.13μg/ml while limits of quantitation were 0.19μg/ml, 0.22μg /ml, for hydrochlorothiazide and candesartan Cilexetil; respectively.

Havaldar and Vairal (2010); developed and validated a RP-HPLC method for the determination of atenolol, hydrochlorothiazide, losartan and valsartan. Separation was achieved with a Nucleolus 100  $C_{18}$  column having 250 x 4.6mm i.d. with 5 µm particle size and potassium dihydrogen phosphate buffer adjusted to pH 3.0 using diluted ortho phosphoric acid and acetonitrile (50:50, v/v) with isocratic program as eluent at a constant flow rate of 1.0/ml. UV detection was performed at 210nm for both.

Joshi et al. (2010); reported HPLC method was developed for simultaneous determination of bisoprolol and hydrochlorothiazide, using  $C_{18}$  column (250 × 4.6mm, 5µm). The mobile phase composed of 0.1M potassium dihydrogen phosphate buffer and acetonitrile (70:30, v/v); using isocratic elution with flow rate of 1ml/min, the eluents were monitored at 228nm. The Linearity ranges of Bisoprolol Fumarate and Hydrochlorothiazide were 2.5- 50μg/ml and 6.25-125μg/ml; Limits of detection of Bisoprolol Fumarate and Hydrochlorothiazide were 0.01 μg/ml and 0.01μg/ml; while limits of quantitation were 0.03μg/ml, 0.05μg/ml respectively.

Kavitha, J. et al. (2011); developed and validated a RP-HPLC method for the analysis of Telmisartan and Hydrochlorothiazide. Princeton SPHER R  $C_{18}$  column was used, the mobile phase was composed of acetonitrile and potassium dihydrogen ortho phosphate (pH 3.5) in ratio 50:50 (v/v), an isocratic elution was used; flow rate was 1ml/min with a total run time of 10 minutes. Sulphadoxine was selected as internal standard. Detection of the multi compounds was carried out at 270nm.

Nalwade et al.(2011); developed and validated RP-HPLC method for the determination of Telmisartan, Amlodipine besylate and Hydrochlorothiazide using C18 column(100  $\times$ 2.1mm, 1.7 $\mu$ m) at 55◦C, using two mobile phases (A) 0.053M sodium perchlorate buffer pH 3.2 and acetonitrile (90:10, v/v) (B) was 0.053M sodium perchlorate buffer pH 3.2 and acetonitrile (20:80,  $v/v$ ); using gradient elution with aflow rate of 0.6 ml/min. Eluents were monitored at 271nm for telmisartan and hydrochlorothiazide, 237nm for Amlodipine. Linearity ranges were  $16.024 - 48.073$  μg/ml,  $2.02 - 6.02\mu$ g/ml and  $5 - 15\mu$ g/ml for of Telmisartan, Amlodipine besylate and Hydrochlorothiazide; respectively.

Swamy et al. (2012); Reported an HPLC-UV method for determination of Aliskiren, Hydrochlorothiazide and amlodipine bisulfate using  $C_{18}$  column (250 × 4.6mm, 5µm) maintained at 25◦C. injection volume was 10μl, mobile phase was Acetonitrile, methanol and phosphate buffer pH 3.0 in the ratio of  $20:50:30 \, (v/v)$ , isocratic elution was used, with 1ml/min flow rate, eluents were monitored at 239 nm. The linearity ranges of Aliskiren, Amlodipine besylate and hydrochlorothiazide were 75-450μg/ml, 5–30μg/ml and 12.5- 75μg/ml, respectively. Limits of detections of Aliskiren, Amlodipine besylate and hydrochlorothiazide were 0.1521μg/ml, 0.2305μg/ml and 0.2132μg/ml, respectively; while limits of quantitation were 0.3015μg/ml, 0.2517μg/ml and 0.2615μg/ml, respectively.

Samya et al. (2012); developed a simultaneous HPLC-UV method for determination of amlodipine, hydrochlorothiazide and valsartan using C18 column (150 x4.6mm, 5μm) at 30<sup> $\degree$ </sup>C, injection volume was 20 $\mu$ L, the mobile phase was phosphate buffer (pH 2.8): acetonitrile (60:40), isocratic elution with a flow rate of 0.8 ml/min. Eluents were monitored at 227 nm. The linearity ranges of amlodipine, hydrochlorothiazide and valsartan were 4μg/ml–28μg/ml, 1μg/ml–12μg/ml, and 5μg/ml–40μg/ml, limits of detection were 1.04μg/ml, 0.39μg/ml and 1.4μg/ml, while limits of quantitation were 3.16μg/ml, 0.81μg/ml and 4.3μg/ml, respectively.

Mamdouh et al. (2012); developed simultaneous HPLC-UV method for determination of valsartan and hydrochlorothiazide, using column  $C_{18}$  (150 x4.6 mm, 5µm), injection volume was 50 μ, the mobile phase was phosphate buffer pH 2.9 acetonitrile and methanol (50:40:10, v/v) using isocratic elution with flow rate 1.4 ml/min and eluents were monitored at 225 nm. Linearity ranges were 12-36 μg/ml and 2-9μg/ml for valsartan and hydrochlorothiazide, respectively; limits of detection and limits of quantitation were not recorded.

Rasha et al. (2013); developed a simultaneous HPLC-UV method for determination of amlodipine besylate, valsartan and hydrochlorothiazide, using  $C_8$  Column (250 x 4.6mm, 5μ) at 25◦C, injection volume was 20μl, the mobile phase was 0.025M phosphoric acid and acetonitrile using gradient elution with 1m/min flow rate. Eluents were monitored at  $Y^{\dagger}$ nm for amlodipine and 225nm for both valsartan and hydrochlorothiazide. The linearity ranges were 5–200μg/ml, 5–200μg/ml and 10–200μg/ml for amlodipine besylate, valsartan and hydrochlorothiazide; respectively. limits of detection were 0.26μg/ml,  $0.24\mu\text{g/ml}$  and  $0.12\mu\text{g/ml}$ , while limits of quantitation were  $0.85\mu\text{g/ml}$ ,  $0.80\mu\text{g/ml}$  and 0.40μg/ml, for amlodipine besylate, valsartan and hydrochlorothiazide; respectively.

Vasanth et al. (2013); reported an HPLC-UV method for simultaneous determination of losartan potassium and hydrochlorothiazide, using  $C_{18}$  column (150 x 4.6mm, 5µm). The mobile phase was orthophosphric acid Buffer pH 3.0 and Acetonitrile in the ratio of 65:35 (v/v), isocratic elution with flow rate 1ml/min was applied, and eluents were monitored at 254nm. The linearity ranges of losartan potassium and hydrochlorothiazide were 2575μg/ml and 25-18.75μg/ml; respectively. Limits of detection were 1.779μg/ml and 0.375μg/ml; while limits of quantitation 5.393μg/ml and 1.138μg/ml for losartan potassium and hydrochlorothiazide, respectively.

Ritesh and Shyam Sunder (2012); developed a simultaneous HPLC-UV method for determination of valsartan, amlodipine and hydrochlorothiazide using  $C_{18}$  column (250  $\times$ 4.6mm, 5μm) at 35◦C. injection volume was 40μL, formic acid buffer pH 3.50, Acetonitrile and Ammonium Formate as mobile phase using gradient elution with 1ml/min flow rate, eluents were monitored at 254. The linearity ranges of valsartan, amlodipine and hydrochlorothiazide were 50–4000ng/ml, 6–200ng/ml and 5-400ng/ml; respectively.

Kumaraswamy et al. (2014); developed a simultaneous HPLC-UV method for determination of Amlodipine, Atenolol and hydrochlorothiazide, using  $C_{18}$  column (250 x 4.6mm, 5μm), at  $20 \pm 1$ °C, injection volume was  $20 \mu$ . The mobile phase was Phosphate buffer pH 6.0, Acetonitrile and Methanol in the ratio of  $30:20:50$  (v/v) and using Isocratic elution with 1ml/min flow rate, eluents were monitored at 240nm. The linearity ranges of Amlodipine, Atenolol and hydrochlorothiazide were 2-12μg/ml, 10-60μg/ml and 2- 12μg/ml; respectively. Limits of detection were 0.0677μg/ml, 0.1379μg/ml and 0.0478μg/ml, while limits of quantitation were 0.2051μg/ml, 0.4180μg/ml and 0.0145μg/ml, for Amlodipine, Atenolol and hydrochlorothiazide, respectively.

Ashutosh et al. (2014); developed RP-HPLC chromatographic method for simultaneous determination of hydrochlorothiazide, Amlodipine and Olmesartan using  $C_{18}$  column (150 x 4.5mm, 5μm) at ambient temperature, injection volume was 20μl, the mobile phase was triethylamine buffer pH 3.5 and Acetonitrile (40:60, v/v) an isocratic elution was applied, 0.8ml/min flow rate used, eluents were monitored at 230nm. The linearity ranges of hydrochlorothiazide, Amlodipine and Olmesartan were 25-62.5μg/ml, 10-25μg/ml and 10-100μg/ml; respectively. Limits of detection were 0.009μg/ml, 0.06μg/ml and 0.06μg/ml; while limits of quantitation  $0.03\mu$ g/ml,  $0.2\mu$ g/ml and  $0.2\mu$ g/ml hydrochlorothiazide, Amlodipine and Olmesartan, respectively.

Ramachandran et al. (2014); a simultaneous HPLC-UV method for determination of Hydrochlorothiazide and Irbesartan was developed using  $C_{18}$  column (250 x 4.6mm, 5µm) at ambient temperature, injection volume was 20μl, the mobile phase was Phosphate buffer pH 6.0 and Acetonitrile in the ratio of  $60:40$  (v/v), using isocratic elution with1ml/min flow rate, eluents were monitored at 235nm for both. The linearity ranges 90-210μg/ml and 7.5-17.5μg/ml for hydrochlorothiazide and Irbesartan; respectively. Limits of detection were 65.19μg/ml and 0.05μg/ml; while limits of quantitation were 197.56μg/ml and 0.15μg/ml for hydrochlorothiazide and Irbesartan; respectively.

Murali et al. (2014); developed simultaneous HPLC/UV method for determination of valsartan and hydrochlorothiazide using column  $C_{18}$  (150 x4.6mm, 5µm), injection volume was 20μl, the mobile phase was phosphate buffer (pH4) and acetonitrile (40:60), using isocratic elution with 1ml/min flow rate. Eluents were monitored at 225 nm. Linearity ranges were 30- 90μg/ml and 300-900μg/ml, limits of detection 2.903 and 2.941μg/ml and limit of quantitation 9.675 and 9.8μg/ml for amlodipine and losartan, respectively.

Renuka, et al. (2016); developed simultaneous HPLC-UV for determination of bisoprolol and hydrochlorothiazide, using  $C_{18}$  column (150 x 4.6mm, 5µm), injection volume was 10μl, and the mobile phase was phosphate buffer pH 2.5 and acetonitrile in the ratio of 80:20 v/v, was selected at flow rate of 1 ml/min. Eluents were monitored at 208nm. The linearity ranges of Bisoprolol and Hydrochlorothiazide were 2.5-75μg/ml, 3-90μg/ml and Limits of detection were 2μg/ml, 0.9μg/ml, while limits of quantitation were 6μg/ml, 1.8μg/ml for bisoprolol and hydrochlorothiazide; respectively.

Raju et al, (2016); validated HPLC-UV for the simultaneous determination of bisoprolol fumarate and hydrochlorothiazide using  $C_{18}$  column (250 x 4.6mm) Injection volume was 20μl, the mobile phase was Acetonitrile and phosphate buffer pH 3.0 in the ratio of 40:60 (v/v); flow rate of 1ml/min. Eluents were monitored at 228 nm. Linearity ranges were 20- 100 μg/ml for both bisoprolol fumarate and hydrochlorothiazide. Limits of detection were 0.544 μg/ml, 0.658 μg/ml, while limits of quantitation were 1.64μg/ml, 1.99 μg/ml bisoprolol fumarate and Hydrochlorothiazide; respectively.

Ravi et al, (2016); reported that HPLC-UV method has been developed and validated for simultaneous determination of bisoprolol and hydrochlorothiazide in its bulk and combined tablet dosage form. Chromatographic separation was  $C_{18}$  column (250 × 4.6mm; 5 $\mu$ m particle size, maintained at a temperature of 30 °C by a mobile phase consisted of 0.1% orthophosphoric acid and acetonitrile  $(55:45, v/v)$  with a flow rate of 1.0 ml/min. The detection wavelength was set at 259 nm for both. The linearity ranges of 40-120μg/ml for bisoprolol and 50-150μg/ml for hydrochlorothiazide. The limit of quantitation was 0.398 and 0.385μg/ml for bisoprolol and hydrochlorothiazide, respectively.

Shakya (2016); developed a simultaneous HPLC-UV method for determination of valsartan and hydrochlorothiazide, using column C18 (150 x 4.6mm, 5μm) maintained at 25◦C, injection volume was  $20\mu L$ , the mobile phase was 0.25 ml/L triethylamine (pH 3.0), methanol and acetonitrile (50:38:37, v/v) using isocratic elution with 1.5 ml/min flow rate. Eluents were monitored at 265 nm. Linearity ranges were 1.25-64.00 μg/ml and 0.195- 10.00 μg/ml, limits of detection were 0.253 and 0.0226 μg/ml, respectively; while limits of quantitation were 0.767 and 0.068 μg/ml for valsartan and hydrochlorothiazide, respectively.

Vatchavai et al. (2017); reported RP-HPLC method for the simultaneous determination of Telmisartan and hydrochlorothiazide in Pharmaceutical formulations. The separation of the drugs was achieved on a  $C_{18}$ , column 250×4.6mm, 5micron size column with a mobile phase consisting of acetonitrile and phosphate buffer pH 3.0 (60:40 v/v) and 1.0 ml/min flow rate, UV detection at 282 nm for both. Linearity was found to be 6-18μg/ml and 6- 18μg/ml for Telmisartan and Hydrochlorothiazide; respectively. Limits of detection were 0.99μg/ml, 1.55μg/ml, while limits of quantitation were 3μg/ml, 4.7μg/ml for Telmisartan and hydrochlorothiazide; respectively.

#### **1.2.2 HPLC/TLC methods**

A HPLC/TLC chromatographic method for determination of losartan potassium and amlodipine was reported (Lakshmi and Lakshmi 2012); usimg chloroform, methanol, acetone and formic acid (7.5: 1.3: 0.5: 0.03, v/v) as mobile phase, eluents were monitored at 254 nm. The RF values amlodipine besylate, hydrochlorothiazide, and losartan potassium were 0.35, 0.57, and 0.74, respectively; linearity ranges were 500–3000 ng per spot for losartan potassium, amlodipine and hydrochlorothiazide.

Savita's Yadav and Janhavirrao (2013); a simultaneous high-performance thin-layer chromatographic method for determination bisoprolol fumarate and hydrochlorothiazide was developed, using precoated silica gel HPTLC aluminum plate 60 F254, ethyl acetate, methanol and ammonia in the ratio of 10:0.5:0.5, (v/v) as mobile phase. Detection was carried out densitometrically at 225nm. The RF value of bisoprolol fumarate and hydrochlorothiazide were 0.60 and 0.38; respectively; linearity ranges were 150- 900ng/spot and 100-600ng/spot for bisoprolol fumarate hydrochlorothiazide; respectively.

#### **1.2.3 A UV-Spectrophotometric methods**

A UV-Spectrophotometric method has been developed for simultaneous estimation of amlodipine besylate and bisoprolol fumarate in combined dosage form was reported Kakde et al. (2008); The method employed simultaneous equation method for analysis using 10% methanol as a solvent. The two wavelengths 222 nm and 365 nm were selected for estimation of bisoprolol fumarate and amlodipine besylate; respectively. Linearity was observed in the concentration ranges of 5–100 μg/ml for both the drugs.

Wankhede et al. (2010); developed a simultaneous spectrophotometric method for the determination of amlodipine besilate, losartan potassium and hydrochlorothiazide using methanol as solvent; the detection wavelengths were 236.5, 254 and 271 nm, respectively, and linearities of procedure were within the concentration ranges  $5-25 \mu g/ml$ ,  $10-50 \mu g/ml$ and 5-25 μg/ml, respectively. They developed also an HPLC method for determination of same drug using column  $C_{18}$  (250x4.6 mm, 5µm) at ambient temperature. The injection volume was 20 μl, the mobile phase was phosphate buffer (pH 3.7): Acetonitrile (57:43), with isocratic elution flow rate of 1.0 ml/min, and eluents were monitored at 232 nm. Linearity ranges were 2-14 μg/ml, 20-140μg/ml and 5-40μg/ml for amlodipine besylate, losartan potassium and hydrochlorothiazide, respectively. Limits of detection and limits of quantitation were not reported for both methods.

A UV-Spectrophotometric method developed for simultaneous estimation of losartan potassium, amlodipine besylate and hydrochlorothiazide in raw materials and in formulations are described was reported (Nagavalli et al. (2010); Calibrations were constructed using the absorption data matrix corresponding to the concentration data

matrix, with measurements in the range of  $230.5$ nm– $350.4$  nm (id = 0.1 nm) in their zero order spectra. The linearity ranges were found to be 8–40 mg/ml, 1–5 mg/ml and 3–15 mg/ml for losartan potassium, amlodipine besylate and hydrochlorothiazide; respectively.

A UV-spectrophotometric method was developed for determination of amlodipine besylate, valsartan and hydrochlorothiazide were reported (Ananda et al. 2011); using methanol: water (1:1) as solvent, the selected wavelengths were 365 nm, 250 nm and 315 nm. The linear concentration ranges were 1μg/ml–32μg/ml, 4μg/ml-40μg/ml and 2μg/ml–  $20\mu\text{g/ml}$ , limits of detection were  $0.2\mu\text{g/ml}$ ,  $0.3\mu\text{g/ml}$  and  $0.25\mu\text{g/ml}$ , while limits of quantitation were 0.55μg/ml, 0.9μg/ml and 0.75μg/ml for amlodipine besylate, valsartan and hydrochlorothiazide respectively.

A UV-Spectrophotometric method was developed for simultaneous estimation of Telmesartan, amlodipine besylate and hydrochlorothiazide in bulk and in combined tablet dosage form was reported (Delhiraj et al. 2012); The method is based on the simultaneous equation method. Telmesartan, amlodipine besylate and hydrochlorothiazide has absorbance maxima at 292.8 nm, 238.5 nm and 271.2 nm; respectively. The linearity ranges of Amlodipine besylate, Telmesartan and hydrochlorothiazide were 4-24μg/ml, 4-  $24\mu\text{g/ml}$  and  $(4-24\mu\text{g/ml})$ ; respectively.

Varsha et al. (2012); reported a simultaneous UV-Spectrophotometric method for determination of amlodipine besylate, hydrochlorothiazide and valsartan using methanol as solvent, the selected wavelengths were 359 nm, 317 nm and 250 nm, the linear concentration ranges were 5μg/ml -25μg/ml, 10μg/ml -50μg/ml and 5μg/ml -25μg/ml, limits of detection were 0.51μg/ml, 0.91μg/ml and 1.57μg/ml, while limits of quantitation were 1.68μg/ml, 3.02μg/ml and 4.77μg/ml respectively.

Sayyed et al. (2015); developed a simultaneous spectrophotometric method for determination of Amlodipine Besylate and Hydrochlorothiazide, using 0.1N sodium hydroxide as solvent and 238nm and 271nm UV detention respectively; linearity of procedure was within the concentration ranges were 5-30μg/ml, 2.5-15μg/ml Amlodipine Besylate and Hydrochlorothiazide, respectively. Limits of detection and limits of quantitation were not recorded.

Bobade and Ganorkar (2017); reported a simultaneous spectrophotometric method for determination of bisoprolol fumarate and Hydrochlorothiazide using 0.1N Sodium hydroxide as solvent. Detection wavelengths were 238.4 and 274 nm. The linarites of their procedure were within the concentration ranges of 8-96μg/ml, 4-48μg/ml for bisoprolol fumarate and hydrochlorothiazide; respectively. Limits of detection and limits of quantitation were not recorded.

#### **1.2.4 UPLC methods**

Anandkumar et al. (2015); a simultaneous UPLC-MS method was developed for determination of amlodipine besilate, losartan potassium and hydrochlorothiazide using  $C_{18}$  column (50 x 2.1 mm, 1.7 µm) at ambient temperature, injection volume was 2µL, the mobile phase was 1% ammonium acetate (pH 2.6) and acetonitrile, gradient elution with 0.4 ml/min flow rate. Eluents were monitored at 254 nm. The linearity ranges of hydrochlorothiazide, amlodipine and losartan were 125ng/ml - 750ng/ml, 50ng/ml - 300ng/ml, and 500ng/ml - 3000ng/ml, limits of detection were 0.6ng/ml, 0.1ng/ml and 2ng/ml, while limits of quantitation were 1ng/ml, 1ng/ml and 5ng/ml, respectively.

Sevinc et al, (2014); validated UPLC/BE chromatographic method for the simultaneous determination of bisoprolol fumarate and hydrochlorothiazide in their combined dosage forms in spiked human urine samples. The separation was achieved on an Acquity UPLC BEH C<sub>18</sub> 1.7<sub>µm</sub> (2.1  $\times$  50 mm) column, at 40 °C with mobile phase of acetonitrile and phosphate buffer pH 3.0, a gradient elution at 225 nm. The linearity ranges 0.5–150μg/ml for hydrochlorothiazide and 0.5–250μg/ml for bisoprolol fumarate. Limit of detection and limit of quantitation for hydrochlorothiazide were 0.01μg/ml and 0.03μg/ml, respectively, and for bisoprolol fumarate were 0.07μg/ml and 0.21μg/ml, respectively.

#### **1.2.5 Other methods**

Bhoya et al. (2013); reported TLC-densitometry method for simultaneous determination of bisoprolol fumarate and hydrochlorothiazide using precoated silica gel HPTLC aluminum plate 60 F254, using chloroform, ethanol and glacial acetic acid in the ratio of 5:1.5:0.2 (v/v) as mobile phase. Detection was carried out densitometrically at 225nm. The  $R_f$  value were bisoprolol fumarate and Hydrochlorothiazide 0.62 and 0.40; respectively.

The linearity ranges were 200–1200ng/spot and 100-800ng/spot for bisoprolol fumarate and Hydrochlorothiazide; respectively.

Bozal et al. (2013); developed a simultaneous voltammetry, chromatographic, and spectrophotometric methods for determination of bisoprolol fumarate and hydrochlorothiazide Differential pulse and square wave voltammetry techniques were used to analyze bisoprolol fumarate and hydrochlorothiazide simultaneously by measuring at about 1400 mv and 1100 mv; respectively. RP-HPLC was the second method for simultaneous analysis of the compounds. The mixture of bisoprolol fumarate, hydrochlorothiazide and Moxifloxacin as an internal standard was separated on  $C_{18}$ column (150  $\times$  4.6mm, 5µm) using acetonitrile and 15 mM phosphate in the ratio of 25: 75.)  $v/v$  as mobile phase with 1ml/min flow rate. The third method was based on first derivative of the ratio spectra method obtained from the measurements of the amplitudes at 246nm, 257nm for bisoprolol fumarate and hydrochlorothiazide, respectively.

## **1.3 Objectives**

- To develop methods for Bisoprolol fumarate and Hydrochlorothiazide.
- To determine System suitability, Specificity, Linearity, Accuracy and Robustness for the developed method.
- To compare the results with acceptance criteria of USP and ICH guidelines.

**Materials and Methods Chapter Two**

# **Chapter Two 2. Materials and Methods**

#### **2.1 Chemicals**

Bisoprolol fumarate (purity: 99.30%), Hydrochlorothiazide (purity: 99.10%) and Tetra butyl ammonium hydroxide 40% were obtained from Aurobindo, Unichem and Emplura India. Methanol (HPLC grade). All other chemicals used were of analytical grade. Purified water from Milli-Q-system (Millipore, Bangalore, India).

#### **2.2 Instruments**

• High Performance Liquid Chromatography HPLC

Type: HPLC prominence – i

Model: LC-2030C3D

Serial No: L21455300660AE

Company: Shimadzu Corporation

Origin: Japan

## • **Analytical Balance**

Type: AY220 Serial No: O4328143000 Capacity: 220 g Readability: 0.1 mg Company: Shimadzu Corporation Origin JAPAN

## • **Ultrasonic**

Model: 621.05.003

Company: ISO Lab Laborgerate -GmbH

Origin: Germany

## • **pH-meter**

 Model: PHS-550 Origin: Romania.

## • **Magnetic Stirrer**

Model: LMS, 1001 Serial No: 2016017862 Company: QAIHAN LAB TECH Co-LTD Origin: Korea

## **2.3 Methods**

#### **2.3.1 Optimized chromatographic conditions**

- Cyanide column ( $250 \times 4.6$ mm, 5 km), and simple isocratic elusion, were used (one pump required) with flow-rate of 0.9ml/min, both, active, ingredients were detected at 228nm, injection volume was 10μl (universal loop) at 30◦C.
- Temperature Procedure was 30°C.

#### **2.3.2 Buffer Solution pH 5.0**

In 1000 ml volumetric flask 980 ml of deionized water, 10 ml of tetra butyl ammonium hydroxide (40%) were mixed, was adjusted pH 5.0 with glacial acetic acid, and the volume was completed to the mark with deionized water.

#### **2.3.3 Mobile Phase**

Mixture of buffer and methanol were prepared in 82:18 v/v ratio. The mixture was shaken, filtered at pump through 0.45μm nylon membrane filter, transferred to solvent reservoir and sonicated for 5 min.

#### **2.3.4 Standard Stock Solution**

0.25g Hydrochlorothiazide and 0.1g bisoprolol fumarate were accurately weighed, transferred quantitatively to the same 100ml volumetric flask, 50ml Methanol were added and sonicated for 5 min, cooled to reach room temperature, and completed to volume with mobile phase.

#### **2.3.5 System Suitability**

Subsequent, dilutions were made from the stock solution with mobile phase to give concentrations of 250μg/ml hydrochlorothiazide and 100μg/ml Bisoprolol Fumarate. System suitability solution was injected six times.

#### **2.3.6 Specificity**

#### • **Standard**

The stock solution was diluted with mobile phase to give the concentrations of 250μg/ml hydrochlorothiazide and 100μg/ml bisoprolol fumarate. System suitability solution was injected six times.

#### • **Placebo**

A placebo equivalent to average weight of one tablet was transferred to 50-ml volumetric flask, the flask was half filled with mobile phase, sonicated for 10 minutes, cooled to room temperature, and the volume was completed to the mark with the same solvent. Subsequent, dilutions were made in mobile phase similar to those made for standard preparation.

#### • **Sample**

Five tablets were placed into a clean, dry 100 ml volumetric flask and dissolved with 10 ml methanol, sonicated for 10 min, cooled, then 50 ml mobile phase was added, sonicated for 20 min, cooled to reach room temperature, and then completed to volume with mobile phase. 5ml were diluted to volume with mobile phase in a 25ml volumetric flask, passed through 0.45μm pore filter.

## • **Sample with fumaric acid Preparation**

10 mg fumaric acid were placed into clean dry 100 ml volumetric flask and dissolved in 10 ml methanol, sonicated for 10 min, cooled, then 50 ml mobile phase were added, sonicated for 20 min, cooled to reach room temperature, and then completed to volume with mobile phase. 1ml was diluted to volume with mobile phase in a 10 ml volumetric flask, passed through 0.45µm pore filter.

## **2.3.7 Linearity**

The stock solution was diluted with mobile phase to give concentrations of 100, 150, 200, 250, 300, 350 and 400μg/ml hydrochlorothiazide and 40.60,80,100,120,140 and 160μg/ml bisoprolol fumarate. Each solution was injected three times and results were collected, LOD and LOQ were calculated from linear regression analysis.

#### **2.3.8 Accuracy**

#### • **Standard**

The stock solution was diluted with mobile phase to give the concentrations of 250μg/ml hydrochlorothiazide and 100μg/ml Bisoprolol Fumarate. System suitability solution was injected six times.

## • **Preparation of Test Solution**

Three 100-ml volumetric flasks were labeled; a placebo equivalent to tablets weight was transferred to each flask. A volume of standard stock solution required to produce 50%, 100%, and 150% tablets content of hydrochlorothiazide and bisoprolol fumarate were placed to flasks. The flasks were half filled with mobile phase, sonicated for 10 minutes, cooled to reach room temperature and completed to the mark with the same solvent. Subsequent dilutions were made with mobile phase like those made for the standard preparation. Each solution was injected three times. The results were collected and subjected to statistical treatments.

## **2.3.9 Precision**

## • **Standard**

The stock solution was diluted with mobile phase to give the concentrations of 250μg/ml hydrochlorothiazide and 100μg/ml bisoprolol Fumarate. System suitability solution was injected six times.

## • **Sample**

Five tablets were taken into clean and dry 100 ml volumetric flask dissolved with 10 ml methanol, sonicated for 10 min, cooled, then 50 ml mobile phase were added, sonicated for 20 min, cooled to reach room temperature, completed to volume with mobile phase. Then 5ml were diluted with mobile phase into 25ml volumetric flask, passed through 0.45μm pore filter.

#### **2.3.10 Robustness**

Accuracy sample solution of target concentration (100%) was used. a method remains unaffected by small, deliberate changes in method parameters like column temperature, flow rate and detection wavelength. Here the column temperature was varied  $\pm 5^{\circ}C$ , detection wavelength varied  $\pm 2$ nm and flow rate was varied  $\pm 0.1$  ml/min.

**Chapter Three Results and Discussions**

## **Chapter Three**

# **3. Results and Discussions**

## **3.1 Results**

## **3.1.1 System Suitability**

Shows Table (3.1) and Table (3.2). System suitability results for bisoprolol fumarate and hydrochlorothiazide; respectively

N <sub>0</sub>	Area	<b>Retention time</b>	<b>Tailing factor</b>	<b>Resolution</b>	<b>Theoretical plates</b>
1	1476536	4.718	1.247	10.468	39604
$\overline{2}$	1477324	4.71	1.241	10.438	39290
3	1475963	4.72	1.238	10.422	39390
$\overline{\mathbf{4}}$	1475068	4.722	1.236	10.421	39316
5	1470496	4.71	1.235	10.385	39211
6	1475461	4.719	1.234	10.358	39092
<b>AVG</b>	1475141.333	4.7165	1.2385	10.41533333	39317.16667
<b>STDEV</b>	2411.26321	0.005205766	0.00484768	0.038913579	173.3717586
RSD%	0.163459809	0.110373498	0.39141541	0.373618186	0.440956898

Table (3.1) System suitability results for bisoprolol fumarate

Table (3.2) System suitability results for hydrochlorothiazide



Shows Table (3.1) and (3.2) Tailing factors, resolution, theoretical plates and %RSD were satisfactory with to USP and ICH guidelines

## **3.1.2 Specificity**

Figure 3.1, Figure 3.2, Figure 3.3 and Figure 3.4 shows the specificity chromatograms for placebo, sample, standard of bisoprolol fumarate, hydrochlorothiazide and Fumaric acid; respectively.







Figure 3.2 **chromatogram for the sample of bisoprolol fumarate and hydrochlorothiazide**



Figure 3.3: **chromatogram for mixed standard of bisoprolol fumarate and hydrochlorothiazide** Shows Figure 3.2 and Figure 3.3 The method was found to be specific to these two active ingredients.



Figure (3.4) **chromatogram for fumaric Acid**

This peak confirmed and given the same retention time as second peak the combination by injection of fumaric acid.

## **3.1.3 Linearity, LOD and LOQ**

## **I. Bisoprolol fumarate**

Shows Table (3.3) linearity results for bisoprolol fumarate

$\frac{6}{6}$	40%	60%	80%	100%	120%	140%	160%
$C. \mu g/ml$	40	60	80	100	120	140	160
$Area - 1$	567911	856268	1123553	1395731	1691940	1973292	2229507
Area $-2$	567682	855959	1123277	1395368	1691052	1968819	2229121
Area $-3$	568128	855616	1121591	1394959	1690548	1977723	2248307
<b>AVG</b> Area	567907	855947.666	1122807	1395352.667	1691180	1973278	2235645
<b>STDEV</b>	223.0269	326.1477	1062.0904	386.2283	704.7723	4452.016	10967.312
$RSD\%$	0.0393	0.0381	0.0946	0.0277	0.0417	0.2256	0.4906

Table (3.3) linearity results for bisoprolol fumarate



Figure 3.5: **Plot of average area versus concentration for bisoprolol fumarate**

Table (3.4) shows regression parameters for Bisoprolol fumarate

<b>Parameter</b>	<b>Value</b>
<b>Regression Coefficient R2</b>	0.999829
<b>Root Mean Squire Error (RME)</b>	8631.25704
Slope $(S)$	13939.728
<b>Intercept</b>	12043.9643

Table (3.4) Regression parameters for data Bisoprolol fumarate

## • **Limit of detection and limit of quantitation for Bisoprolol fumarate**

LOD =3\*RMSE/S

 $= 3* 8631.257/13939.728 = 1.857 \mu g/ml$ 

LOQ=10\*RMSE/S

=10\*8631.257/13939.728=6.192μg/ml

#### **II. Hydrochlorothiazide**

Shows table (3.5) linearity results for Hydrochlorothiazide

$\frac{6}{9}$	40%	60%	80%	100%	120%	140%	160%
$C. \mu g/ml$	100	150	200	250	300	350	400
$Area - 1$	4671998	6863543	9069697	11397175	13575926	15737025	17890700
$Area - 2$	4646667	6862508	9086374	11395711	13581165	15679591	17893591
$Area - 3$	4670935	6863214	9045076	11367125	13551764	15716988	17865704
<b>AVG</b> Area	4663200	6863088	9067049	11386670	13569618	15711201	17883331
<b>STDEV</b>	14327.859	528.81975	20775.951	16942.57552	15682.62396	29150.9900	15334.2898
$RSD\%$	0.3072538	0.0077052	0.2291369	0.148793063	0.115571592	0.18554271	0.08574627

Table (3.5) linearity result for Hydrochlorothiazide



Figure 3.6: **Plot of average area versus concentration for hydrochlorothiazide** The concentrations range was (40–160) μg/ml for bisoprolol fumarate and (100–400) μg/ml for hydrochlorothiazide. Each solution was injected in triplicate. Plot of average area versus prepared concentrations, correlation  $(R^2 = 0.999)$  for both components indicates a very good linearity.

Shows Table (3.6) Regression parameters for Hydrochlorothiazide by excel program





• **Limit of detection and limit of quantitation for Hydrochlorothiazide**

 $LOD = 3*RMSE/S$  $= 3* 52699.89639/44185.13595 = 3.578 \mu g/ml$ LOQ=10\*RMSE/S =10\*52699.89639/44185.13595=11.927μg/ml

#### **3.1.4 Accuracy**

Table (3.7) shows the results of mixed standard of bisoprolol fumarate and hydrochlorothiazide while the accuracy results for bisoprolol fumarate and hydrochlorothiazide samples were shown in Table (3.8) and Table (3.9), respectively; summary of accuracy results for both components is shown in Table  $(3.10)$ .



Table (3.7) Results of hydrochlorothiazide and bisoprolol fumarate standard for accuracy test

Table (3.8) Accuracy results for bisoprolol fumarate

<b>Content</b>	50%	100%	150%
1	726795	1415612	2169404
$\overline{2}$	726784	1416273	2141343
$\overline{\mathbf{3}}$	725474	1415666	2169404
AVG	726351	1415850	2160050
<b>STDEV</b>	759.524	367.035	16201.026
$RSD\%$	0.10457	0.02592	0.75003
<b>Recovery</b>	51.371	100.135	152.768
Recovery%	102.741	100.135	101.845

Table (3.9) Accuracy results for hydrochlorothiazide

<b>Content</b>	50%	100%	150%
1	6044602	12026875	18223119
$\overline{2}$	6024101	12033751	18290031
3	6045482	12028800	18316606
<b>AVG</b>	6038062	12029809	18276585
<b>STDEV</b>	12098.296	3547.238	48172.028
RSD%	0.20037	0.02949	0.26357
<b>Recovery</b>	50.271	100.157	152.166
Recovery%	100.543	100.157	101.444

Table (3.10) Summary of accuracy results for Recovery % bisoprolol fumarate and hydrochlorothiazide



The recovery percentage for bisoprolol fumarate at the above concentrations was found to be 102.7, 100.1 and 101.9, respectively; while for hydrochlorothiazide, it was 100.5, 100.1 and 101.4 respectively. The average of recovery percentage for bisoprolol fumarate and hydrochlorothiazide was 101.5% and 100.7%, respectively, (this results within limits) indicates that the proposed method is accurate.

## **3.1.5 Precision**

## **I. Intraday Precision**

Table (3.11) shows results of bisoprolol fumarate and hydrochlorothiazide mixed standard for intraday precision test.

Table (3.11) for the results of bisoprolol fumarate and hydrochlorothiazide mixed standard for intraday precision test



Table (3.12) and Table (3.13) show intraday precision for bisoprolol fumarate and hydrochlorothiazide, respectively





N <sub>O</sub>	1 <sup>st</sup>	2 <sub>nd</sub>	3rd	4 <sup>th</sup>
1st trial	12405489	12311249	12376373	12316188
2nd trial	12456555	12358094	12358972	12349591
3rd trial	12460101	12363895	12312891	12350422
<b>AVG</b>	12440715	12344412.67	12349412	12338733.67
<b>STDEV</b>	30558.1	28866.67	32802.99	19529.5
RSD%	0.24563	0.233844	0.265624	0.15828
<b>Recovery</b>	100.72987	99.950126	99.990605	99.904145
Recovery%	100.72987	99.950126	99.990605	99.904145

Table (3.13) Intraday results for hydrochlorothiazide

Table (3.14) shows Summary of intraday precession for bisoprolol fumarate and hydrochlorothiazide

Table (3.14) Summery intraday precession for bisoprolol fumarate and hydrochlorothiazide

<b>Trial</b>	<b>Bisoprolol fumarate</b>	Hydrochlorothiazide
1 <sup>st</sup>	102.3195	100.7299
2nd	101.76896	99.95013
3rd	101.9461	99.99061
4th	101.9145	99.90414
<b>AVG</b>	101.987265	100.143695
<b>STDEV</b>	0.234539599	0.392396622
RSD%	0.229969496	0.391833577

Bisoprolol fumarate intraday precision, the RSD for the recovery percentage of assay repetitions was 0.025%, 0.40%, 0.211% and 0.18% respectively; whereas for hydrochlorothiazide RSD was 0.245%, 0.233%, 0.265% and 0.158%; respectively. The RSD values was found to be less than 2.0% so it is acceptable according to USP and ICH guidelines.

#### **II. Interday Precision**

Table (3.15) shows results of bisoprolol fumarate and hydrochlorothiazide mixed standard for interday precision test

Table (3.15) results of bisoprolol fumarate and hydrochlorothiazide mixed standard for interday precision test



Table (3.16) shows intraday precision for both components, respectively. Table (3.11) shows the summary of interday precision





Shows table (3.17) shows interday precision summary for both bisoprolol fumarate and hydrochlorothiazide





For the interday, the RSD for the recovery percentage of bisoprolol fumarate three assay repetitions was 0.106%, 0.024% and 0.066%; respectively, whereas for hydrochlorothiazide RSD was 0.059%, 0.005% and 0.20%, respectively. The RSD values was found to be less than 2.0% so it is acceptable according to USP and ICH.

## **3.1.6 Robustness**

The robustness of the method was determined as per ICH guidelines under a variety of conditions like change in Temperature, wavelength and flow rate. The results obtained by deliberately variation in method parameters.

## **3.1.6.1 Robustness study of Bisoprolol fumarate**

#### **i) Optimized conditions**

Shows table (3.18) results of bisoprolol fumarate sample at optimum conditions

<b>Trial No</b>	Area	<b>Retention Time</b>	<b>Theoretical plates</b>	<b>Tailing factor</b>	<b>Resolution</b>
1	1410282	4.935	41371	1.195	9.64
$\overline{2}$	1410818	4.938	41649	1.194	9.648
3	1411423	4.937	41641	1.194	9.651
<b>AVG</b>	1410841	4.936667	41553.67	1.194333	9.646333
<b>STDEV</b>	570.8476154	0.001528	158.2445	0.000577	0.005686
$RSD\%$	0.040461513	0.030942	0.38082	0.048341	0.058947

Table (3.18) results of optimum conditions of bisoprolol fumarate

#### **ii) 5 ◦C more**

Shows table (3.19) results of bisoprolol fumarate at 35◦C column temperature

<b>Trial No</b>	Area	<b>Retention Time</b>	<b>Theoretical plates</b>	<b>Tailing factor</b>	<b>Resolution</b>
	1466463	4.846	43990	1.197	8.668
$\overline{2}$	1465325	4.847	44157	1.198	8.663
3	1466023	4.847	44141	1.2	8.661
AVG	1465937	4.846667	44096	1.198333	8.664
<b>STDEV</b>	573.8536399	0.000577	92.14662	0.001528	0.003606
$RSD\%$	0.03914586	0.011912	0.208968	0.127471	0.041615

Table (3.19) results of bisoprolol fumarate at 35◦C column temperature

#### **iii) 5 ◦C less**

Shows table (3.20) results of bisoprolol fumarate at 25◦C column temperature

<b>Trial No</b>	Area	<b>Retention Time</b>	<b>Theoretical plates</b>	<b>Tailing factor</b>	<b>Resolution</b>
	1409363	4.942	41291	1.194	9.702
$\overline{2}$	1411298	4.935	41269	1.196	9.631
3	1412089	4.936	41256	1.197	9.619
AVG	1410916.667	4.937667	41272	1.195667	9.650667
<b>STDEV</b>	1402.437283	0.003786	17.69181	0.001528	0.044859
RSD %	0.099399016	0.076675	0.042866	0.127755	0.464828

Table (3.20) results of bisoprolol fumarate at 25◦C column temperature

#### **iv) 10 % more Flow rate**

Shows table (3.21) results of bisoprolol fumarate at 1ml/min flow rate

Table (3.21) results of bisoprolol fumarate at 1ml/min flow rate



#### **v) 10 % less Flow rate**

Shows table (3.22) results of bisoprolol fumarate at 0.8ml/min flow rate

<b>Trial No</b>	Area	<b>Retention Time</b>	<b>Theoretical plates</b>	<b>Tailing factor</b>	<b>Resolution</b>
	1588774	5.549	44867	1.191	9.99
$\overline{2}$	1588679	5.551	44957	1.192	9.997
3	1589747	5.553	45043	1.193	9.999
<b>AVG</b>	1589066.667	5.551	44955.67	1.192	9.995333
<b>STDEV</b>	591.0975667	0.002	88.00758	0.001	0.004726
$RSD\%$	0.037197783	0.03603	0.195765	0.083893	0.04728

Table (3.22) results of bisoprolol fumarate at 0.8ml/min flow rate

#### **vi) 2nm more Wavelength**

Shows table (3.23) results of bisoprolol fumarate at 230nm wavelength

Table (3.23) results of bisoprolol fumarate absorbance at 230nm wavelength



#### **vii) 2nm less Wavelength**

Shows table (3.24) results of bisoprolol fumarate at 226nm wavelength

Table (3.24) results of bisoprolol fumarate absorbance at 226nm wavelength



## **3.1.6.2 Robustness study of hydrochlorothiazide**

## **i) Optimized conditions**

Shows table (3.25) results of hydrochlorothiazide sample at optimum conditions



Table (3.25) results of hydrochlorothiazide sample at optimum conditions

#### **ii)5 ◦C more**

Shows table (3.26) results of hydrochlorothiazide at 35◦C column temperature

Table (3.26) results of hydrochlorothiazide at 35◦C column temperature



#### **iii) 5 ◦C less**

Shows table (3.27) results of hydrochlorothiazide at 25◦C column temperature

Table (3.27) results of hydrochlorothiazide at 25◦C column temperature



#### **iv) 10 % more flow rate**

Shows table (3.28) results of hydrochlorothiazide at 1ml/min flow rate

<b>Trial No</b>	<b>Area</b>	<b>Retention Time</b>	<b>Theoretical plates</b>	<b>Tailing factor</b>	<b>Resolution</b>
	10800235	6.996	53330	1.156	9.299
	10792695	6.996	53335	1.156	9.297
3	10790233	6.993	53386	1.155	9.297
<b>AVG</b>	10794387.67	6.995	53350.33	1.155667	9.297667
<b>STDEV</b>	5211.414523	0.001732	30.98925	0.000577	0.001155
RSD %	0.048278927	0.024761	0.058086	0.049958	0.012419

Table (3.28) results of hydrochlorothiazide at 1ml/min flow rate

#### **v) 10 % less flow rate**

Shows table (3.29) results of hydrochlorothiazide at 0.8ml/min flow rate

<b>Trial No</b>	Area	<b>Retention Time</b>	<b>Theoretical plates</b>	<b>Tailing factor</b>	<b>Resolution</b>
	13409613	8.747	59746	1.15	9.99
2	13406495	8.748	59887	1.15	9.997
3	13413274	8.748	59982	1.15	9.999
<b>AVG</b>	13409794	8.747667	59871.67	1.15	9.995333
<b>STDEV</b>	3393.122603	0.000577	118.7448	$\overline{0}$	0.004726
$RSD\%$	0.025303316	0.0066	0.198332	$\Omega$	0.04728

Table (3.29) results of hydrochlorothiazide at 0.8ml/min flow rate

#### **vi) 2nm more Wavelength**

Shows table (3.30) results of hydrochlorothiazide absorbance at 230nm wavelength

<b>Trial No</b>	Area	<b>Retention Time</b>	<b>Theoretical plates</b>	<b>Tailing factor</b>	<b>Resolution</b>
	9667692	7.774	55928	1.156	9.62
$\mathbf{2}$	9666205	7.775	56005	1.156	9.628
$\mathbf{3}$	9669871	7.773	56085	1.157	9.63
<b>AVG</b>	9667922.66	7.774	56006	1.156333	9.626
<b>STDEV</b>	1843.8531	0.001	78.5047	0.0005	0.0052
$RSD\%$	0.0190	0.0128	0.140	0.049	0.054

Table (3.30 results of hydrochlorothiazide absorbance at 230nm wavelength

#### **vii) 2nm less Wavelength**

Shows table (3.31) results of hydrochlorothiazide absorbance at 226nm wavelength

<b>Trial No</b>	Area	<b>Retention Time</b>	<b>Theoretical plates</b>	<b>Tailing factor</b>	<b>Resolution</b>
	13503899	7.775	56647	1.15	9.661
$\mathbf{2}$	13500076	7.776	56720	1.151	9.669
$\overline{3}$	13503510	7.774	56796	1.151	9.671
<b>AVG</b>	13502495	7.775	56721	1.150667	9.667
<b>STDEV</b>	2103.925141	0.001	74.50503	0.000577	0.005292
$RSD\%$	0.015	0.012	0.131	0.050	0.054

Table (3.31) results of hydrochlorothiazide absorbance at 226nm wavelength

The robustness of the method was assessed by assaying test solutions under different analytical conditions deliberately changed from the original conditions such as column temperature ( $\pm$  5°C), flow rate ( $\pm$  0.1ml) and detection wavelength ( $\pm$ 2 nm). RSD for the area at all different conditions for target indicates that the proposed method is robust.

#### **3.2 Discussions**

System suitability was carried out with six injections of solutions of 100% concentration having 100μg/ml of Bisoprolol Fumarate and 250μg/ml Hydrochlorothiazide of each in to the chromatographic system. Since the number of theoretical plates for the two drugs was above 2000, it indicates that the column was efficient in separating all the three drugs. Tailing factors, resolution and %RSD were satisfactory with to USP and ICH guidelines. Linearity of this method was checked using seven solutions centered with the target concentration, the concentrations range was  $(40-160)$  μg/ml for bisoprolol fumarate and (100–400) μg/ml for hydrochlorothiazide. Each solution was injected in triplicate. Plot of average area versus prepared concentrations indicates a very good linearity correlation  $(R<sup>2</sup>)$ =0.999) for both components. The limit of detection for bisoprolol fumarate and hydrochlorothiazide was found to be 1.8575μg/ml and 3.57811μg/ml; respectively, whereas the limit of quantitation was found to be 6.19184 μg/ml and 11.92μg/ml, respectively.

In specificity tests, none of placebo peaks had same retention time of active ingredients peaks, this indicates that the excipients used in the formulation did not interfere in the estimation when we used this method for assay in tablets. The second peak in Sample and standard was confirmed as fumarate by injection of fumaric acid alone and given the same retention time as the combination. Shows (Figure 3.2, 3.3 and Figure 3.4).

Accuracy was evaluated for bisoprolol fumarate and hydrochlorothiazide using three concentrations in content of 50%, 100%, and 150 of target concentration. The recovery percentage for bisoprolol fumarate at the above concentrations was found to be 102.741, 100.135 and 101.845, respectively; while for hydrochlorothiazide, it was 100.543, 100.157 and 101.444 respectively. The average of recovery percentage for bisoprolol fumarate and hydrochlorothiazide was 101.5736% and 100.7146%, respectively, (this results within limits) indicates that the proposed method is accurate.

The precision of the methods was examined by estimating the corresponding recovery percentages four times on the same day in intraday precision and three times at three different days for inter day precision. The concentrations used was 100% of target concentration as per ICH. For bisoprolol fumarate intraday precision, the RSD for the recovery percentage of assay repetitions was 0.025%, 0.40%, 0.211% and 0.18% respectively; whereas for hydrochlorothiazide RSD was 0.245%, 0.233%, 0.265% and 0.158%; respectively. For the interlay, the RSD% for the recovery percentage of bisoprolol fumarate three assay repetitions was 0.106%, 0.024% and 0.066%; respectively, whereas for hydrochlorothiazide RSD was 0.059%, 0.005% and 0.20%, respectively. The RSD values was found to be less than 2.0% so it is acceptable according to USP and ICH. The robustness of the method was assessed by assaying test solutions under different analytical conditions deliberately changed from the original conditions such as column temperature  $(\pm 5^{\circ}C)$ , flow rate ( $\pm 0.1$ ml) and detection wavelength ( $\pm 2$  nm). RSD for the area at all different conditions for target indicates that the proposed method is robust.

#### **3.3 Conclusions**

- The proposed method is simple, sensitive and reproducible.
- Statistical analysis of the results has been carried out revealing high accuracy and good precision.
- The RSD for all parameters was found to be less than two, which indicates the validity of method and assay results obtained by this method are in fair agreement with USP and ICH guidelines.
- The developed method can be used for routine quantitative simultaneous estimation of bisoprolol fumarate and hydrochlorothiazide in multicomponent pharmaceutical preparation.

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**Appendixes**

## **Published Papers**

Eltoum M. S. A and Elhory, S.E.H. (2018), Validation methods for a simultaneous determination of bisoprolol fumarate and hydrochlorothiazide multi-component products, examining system suitability, specificity, linearity and accuracy. *International Journal of Research in Pharmacy and Pharmaceutical Sciences*, **3**(1),172-179.

Eltoum M. S. A and Elhory, S.E.H. (2018), Validation methods for a simultaneous determination of bisoprolol fumarate and hydrochlorothiazide multi-component products, examining system suitability, precision and robustness. *International Journal of Research in Pharmacy and Pharmaceutical Sciences*, **3**(1),236-243.

#### Paper NO.1



## Paper NO.2

